X

International Journal of Engineering Sciences & Research Technology

(A Peer Reviewed Online Journal) Impact Factor: 5.164

Chief Editor Executive Editor Dr. J.B. Helonde Mr. Somil Mayur Shah

IJESRT INTERNATIONAL JOURNAL OF ENGINEERING SCIENCES & RESEARCH TECHNOLOGY

CHEMICAL PROFILING AND ADME PREDICTION OF SARACA ASOCA (ROXB)

WILDE

Shweta R. Gophane*1 & C.N.Khobragade²

*1&2School of Life Sciences, Swami Ramanand Teerth Marathwada University, Nanded (MS)-431606

DOI: 10.5281/zenodo.3778234

ABSTRACT

The dried powdered bark of *Saraca asoca* was extracted using ethyl acetate, acetone, ethanol and distilled water by a three-step sequential extraction procedure. Among all extracts, ethanolic extract showed highest phenol (438.78±0.86 mg/g gallic acid equivalent), proanthocyanidins (0.8±0.05) and coumarin **(**1.192±0.2 mg/g) content while highest flavonoid $(179.5\pm0.47 \text{ mg/g}$ rutin equivalent) were detected in ethyl acetate extract. Highest DPPH free radical quenching activity was observed in ethanolic extract of *Saraca asoca* (IC₅₀ value 48.43) which was significantly comparable with ascorbic acid (IC₅₀ value 14.12). The UV-visible and FTIR spectrum confirmed the presence of alcohols, phenols, alkanes, alkyl halides, aldehydes, carboxylic acids, aromatics, nitro compounds and amines in ethanolic extract. The FTIR, GC-MS spectrum profile along with the (ADME)-Toxicity analysis unveiled that the medicinally important plant *Saraca asoca* bark extract having various bioactive compounds and can be used as therapeutic agents.

KEYWORDS: ADMET, FTIR, GC-MS, *Saraca asoca*.

1. INTRODUCTION

Saraca asoca a component of several ayurvedic and traditional ethnomedical preparations belongs to the family *Caesalpinaceae* which is commonly known as ashoka meaning that "without sorrow" or which that gives no grief. In India, Ashoka is used in many pharmacological activities like anti-cancer, anti menorrhagic, anti oxytoxic, anti –microbial activity and have extend uses in ayurveda, unani and homeopathy [1]. It has many uses in treatment of skin infections, CNS function and genitor-urinary functions. It's bark mostly beneficial in uterine inertia, uterine pain, urinary calculus, dysurea, Leucorrhoea and urinary stone [2]. Some of the most important bioactive phytochemical constituents such as glycosides, alkaloids, flavonoids, tannins, steroids, terpenoids, essential oils and phenolic compounds are found in *Saraca asoca*[3]. Phenolics and flavonoids are polyphenols, an important class of secondary plant metabolites possessing an impressive array of pharmacological activity which include free radical scavenging, inhibition of hydrolytic and oxidative enzymes and anti-inflammatory action etc [4]. Oxidation reaction can produce free radicals, which can damage cells and tissues. Recently, there has been an upsurge of interest in the therapeutic potential of medicinal plants as antioxidants in reducing such free radical-induced tissue injury. As a result, profiling of medicinal plant by some platform technology such as gas chromatography-mass spectrometer (GC-MS), liquid chromatography-mass spectrometer as well as it's holistic overview of all metabolites present in extracts became crucial and valuable. Spectroscopic (UV-Vis, FTIR) methods together or separate can be used because of its simplicity, cost-effective and rapid test for detecting phytocomponents. Hence the present study is designed to evaluate the phytochemical profile of *Saraca asoca*, with the aid of GC-MS, UV-VIS and FTIR Techniques, to ascertain the rationale for its use in traditional medicine.

2. MATERIALS AND METHODS

2.1 Drugs

Saraca asoca barks were procured from Yogesh pharma Pvt. Ltd., Nanded (MS), India. It was identified and authenticated at Department of Botany, Babasaheb Ambedkar Marathwada University Aurangabad, India. The barks were washed thoroughly with water to remove dust and dried under the shade at room temperature for 5 days. The dried barks were ground using blender to obtain the course powder and kept in an air tight container

http: // www.ijesrt.com**©** *International Journal of Engineering Sciences & Research Technology* $\lceil 1 \rceil$

ISSN: 2277-9655

[Gophane, *et al.,* **9(4): April, 2020] Impact Factor: 5.164**

ISSN: 2277-9655 IC[™] Value: 3.00 CODEN: IJESS7

till further use. Powdered plant material (150g) were successively extracted using Soxhlet apparatus using the solvents in order of increasing polarity viz., C), ethyl acetate, acetone, ethanol and distilled water. Each time the marc was dried and later extracted with other solvents. All the extracts were concentrated by distilling the solvent in a rotary vacuum evaporator.

2.2 Phytochemical Screening of Bark Extract

The four extracts obtained from the powdered bark of *Saraca asoca* were subjected to phytochemical tests to investigate the presence or absence of active secondary metabolites using standard procedures [5].

2.3 Qntitative Estimation of Total Phenols

Folin*-*Ciocalteu reagent was used to determine the total phenolic content (TPC) of the various organic crude extracts with slight modification [6]. 1.5 ml Folin-Ciocalteu reagent (diluted 1:10 with de*-*ionized water) was added to the 0.5 mL of the plant extract sample, and was vortexed for 5 min, followed by addition of 3 mL of sodium carbonate solution (7.5%). This reaction mixture was incubated for 30 min at room temperature in dark. The absorbance of the resulting blue colour was measured by using double beam UV*-*Vis spectrophotometer (UV Shimadzu*-* 1800) at fixed wavelength of 760 nm. The calibration curve was prepared by employing gallic acid at concentrations of 10 to100 μg/ml. The TPCs were determined using linear regression equation obtained from the standard plot of gallic acid. The content of total phenolic compounds was calculated as mean $\pm SD$ $(n=3)$ and expressed as mg/g gallic acid equivalent (GAE) of dry extract.

2.4 Quantitative Estimation of Total Flavonoid

Aluminum chloride colorimetric method was used for flavonoids determination as described by Madaan *et al.*[7]. 0.5 mL of each plant extract was mixed with 1.5 mL of ethanol, 0.1 mL of 10% aluminum chloride, 0.1 mL of 1 M potassium acetate and 2.8 mL of distilled water. The reaction mixture was allowed to stand at room temperature for 30 min and the absorbance of the reaction mixture was measured at 415 nm. The calibration curve was prepared by using rutin at concentrations of 10 to100 μg/ml in ethanol. The amount of flavonoid was calculated from linear regression equation obtained from the rutin calibration curve. The flavonoid content was calculated as mean \pm SD ($n=3$) and expressed as mg/g of rutin equivalent (RE) of dry extract.

2.5 Quantitative Estimation of total Proanthocyanidins

The total proanthocyanidin content were estimated using the procedure reported by Sun *et al*[8].A volume of 0.5 mL of 0.1 mg/mL of extract solution was mixed with 3.0 mL of 4% vanillin-methanol solution and 1.5 mL hydrochloric acid, the mixture was allowed to stand for 15 min at room temperature, the absorbance was measured at 500 nm. The proanthocyanidins content was calculated as mean±SD (*n=*3) and expressed as mg/g of catechin equivalent of dry extract.

2.6 Quantitative Estimation of total Coumarin

Coumarin content was estimated with slight modification in procedure reported by Bruna Medeiros-Neves *et al*[9].A volume of 0.5 mL of the extract was mixed with 2 ml of distilled water and 0.5 mL of lead acetate solution. The sample was shaken and then 7 ml of distilled water were added before transferring 2mL of this solution to a new test tube and added 8 mL of hydrochloric acid solution. The sample was incubated at room temperature for 30 minutes and the absorbance was measured at 500 nm. The coumarin content was calculated as mean±SD (*n=*3) and expressed as mg/g of coumarin equivalent of dry extract.

2.7 Free Radical Scavenging Activity by DPPH Assay Method

The capacity of the extracts and ascorbic acid to scavenge the stable 1, 1-diphenyl-2-picrylhydrazyl (DPPH) free radical was measured as per stated method [10]. 0.1 mL of extracts and ascorbic acid of different concentrations (10to200 μg/ml) was mixed with 2.9 mL of 0.1mM DPPH solution. The solution was rapidly mixed and allowed to stand in dark at room temperature for 30 min. The blank was prepared in a similar way without extract or ascorbic acid. The decrease in absorbance of each solution was measured at 517 nm using UV*-*Vis spectrophotometer. Negative control was prepared by mixing 0.1 mL of ethanol with 2.9 mL of DPPH solution. The percentage of radical scavenging activity of tested extracts and positive control ascorbic acid was calculated by using the following formula:

[Gophane, *et al.,* **9(4): April, 2020] Impact Factor: 5.164 IC[™] Value: 3.00 CODEN: IJESS7** Free radical scavenging activity $(\%) =$ [Ac-As] ×100 A_C

Where Ac=Absorbance of control and As=Absorbance of sampleat 517 nm.

The concentration of sample required to scavenge 50% of DPPH free radical (IC_{50}) was determined from the curve of percent inhibitions plotted against the respective concentration.

2.8 UV-VIS and FTIR Spectroscopic analysis

The ethanolic extract was examined under visible and UV light for proximate analysis. For UV-VIS and FTIR spectrophotometer analysis, the extract was centrifuged at 5000 rpm for 15 min and filtered through Whatman No. 1 filter paper. The sample is diluted to 1:10 with the same solvent. The extract was scanned in the wavelength ranging from 200-1100 nm using UV-1800 Shimadzu UV Spectrophotometer and the characteristic peaks were detected. FTIR analysis was performed using Thermofisher Scientific, model-Nicolet 6700 FT-IR system in a scan range of 400 -4000 cm⁻¹and characteristic peaks and their functional groups were detected.

2.9 Gas Chromatography-Mass Spectrometry Analysis

The GC-MS analysis of the ethanolic extract was carried out using a Agilent 7890 A gas chromatogram equipped and coupled to a mass detector 5975 MSD spectrometer with DB 5 MS and $30m \times 0.25$ µm DF of capillary column. Ultra*-*high purity helium (99.99%) was used as carrier gas at a constant flow rate of 1.0 mL/min. The injection, transfer line and ion source temperatures were all 290 °C. The ionizing energy was 70 eV. Electron multiplier voltage was obtained from autotune. The oven temperature was programmed from 60 °C (hold for 2 min) to 320 °C at a rate of 3 °C/min. The crude ethanolic extract of *Saraca asoca* was diluted with appropriate solvent (1/10, V/V) and filtered. The particle*-*free diluted crude extracts (1 μL) were taken in a syringe and injected into injector with a split ratio 30:1. All data were obtained by collecting the full*-*scan mass spectra within the scan range 30*-*600 amu. The percentage composition of the crude extract constituents was expressed as a percentage by peak area. The identification and characterization of chemical compounds in ethanolic crude extract was based on GC retention time. AMDIS and NIST Version-Year 2011 was used MS data library and comparing the spectrum obtained through GC-MS compounds present in the plants sample was identified.

2.10ADMET predictions

ADMET (Absorption, Distribution, Metabolism, Excretion and Toxicity) analyses constitutes the pharmacokinetics of a drug molecule [11]. In this study, prediction and significant descriptors of drug likeness such as mutagenicity, toxicological dosage level and pharmacologically relevant properties of the compounds were predicted using Swissadme (http://www.swissadme.ch) and admetSAR (lmmd.ecust.edu.cn:8000) servers.

2.11Data Analysis

The results are expressed as mean ± SD. Student's *t-*test and one way ANOVA were applicable and used to analyze level of statistical significance between groups. *P<*0.05 were considered statistically significant. Linear regression analysis was used to calculate the IC_{50} values.

3 RESULTS AND DISCUSSION

3.1 Preliminary Phytochemical Screening

Preliminary phytochemical analysis showed the presence of major classes of secondary metabolites such as tannins, saponins, flavonoids, cardiac glycosides, *etc*. in extracts [Table I]. Alkoids were found to be absent in all the extracts.

3.2 Quantitative Estimation of phytoconstituents

The present study revealed relatively good amount of phenols, flavonoids, coumarin and proanthocyanidins contents in *Saraca asoca* extracts as shown in (Table II). The correlation coefficient of total phenols, total flavonoids, coumarin and total proanthocyanidins were $(r=0.993, 0.984, 0.992,$ and 0.970) respectively (Fig. 1).

ISSN: 2277-9655

3.3 Estimation of Free Radical Scavenging Activity

The antioxidant activity of *Saraca asoca* was evaluated using the DPPH free radical scavenging method. Ascorbic acid was used as standard compound. The *Saraca asoca* exhibited strong antioxidant activity in the DPPH inhibition assay as evidenced by the low $IC₅₀$ values and were compared with standard antioxidant, ascorbic acid. All the extracts showed a dose dependent scavenging activity of DPPH comparable to standard antioxidant (Fig. 2). The IC₅₀ value (in $\mu g/mL$) of the extracts was found in the order of distilled water>acetone>ethyl acetate>ethanol extract (Table III). IC₅₀ value of ascorbic acid (14.12μg/mL) was less than the ethanol extract (18.4μg/mL).Higher phenol and flavonoid content in the plant tissue can also lead to increase in antioxidant activity [12].

3.4 Spectrophotometric analysis

The UV-VIS profile of *Saraca asoca* extract was selected at the wavelength of 200 to 400 nm and the spectrum profile showed the peaks at 344 and 232 nm with the absorption 3.600 and 1.256 respectively (Fig.3; Table IV). Occurrence of peaks at 234-676 nm reveals the presents of phenolic and alkaloids.The FTIR spectrum was used to identify the functional group of the active components based on the peak value in the region of infrared radiation. The results of FTIR peak values and functional groups were represented in (Fig. 4; Table V). FTIR spectrum confirmed the presence of alcohols, phenols, alkanes, alkynes, alkyl halides, aldehydes, carboxylic acids, aromatics, nitro compounds and amines in ethanol extract. On comparison of the spectra of bark shows that the extract has some similar alkaloid, phenolics and glycosides compounds[13]. Hence, UV and FTIR spectroscopy is reliable and sensitive method for detection of bioactive constituents.

3.5 Chemical Composition of Extracts by GC-MS Analysis

GC-MS chromatogram of the ethanolic extract of *S.asoca* (Fig. 5) showed 14 peaks indicating the presence of 14 phytochemical constituents. All the constituents were characterized and identified by comparison of the mass spectra of the constituents with the AMDIS and NIST Version-Year 2011library. The active principles with their retention time (RT), molecular formula, molecular weight (MW) and concentration (%) are presented in (Table VI). The prevailing compounds were à-D-(+)-Talopyranose, pentakis(trimethylsilyl) ether (68%) followed by Sorbopyranose, 1,2,3,4,5-pentakis-O-(trimethylsilyl)-, L-(61%), D-Pinitol, pentakis(trimethylsilyl) ether (57%) ,Mannose, 6-deoxy-2,3,4,5-tetrakis-O-(trimethylsilyl)-, L- (56%), Myo-Inositol, 1,2,3,4,5,6 hexakis-O-(trimethylsilyl) (55%), , >Arabinopyranose, tetrakis-O-(trimethylsilyl)-, à-D- (47%) and Mannose, 2,3,4,5,6-pentakis-O-(trimethylsilyl)-, D-(37%). The biological activities listed (Table VII) are based on Dr.Duke's Phytochemical and Ethnobotanical Databases by Dr. Duke of the Agricultural Research Service/USDA [14]. The presence of various bioactive compounds in *Saraca asoca* justifies the use of the bark for various ailments by traditional practitioners.

3.6 ADMET Predictions

The potential ADME profiles of the compounds as predicted using admetSAR server, while the distribution profile of the compounds as obtained from admetSAR server is shown in (Table VIII). Computational study for the prediction of the relevant properties influencing bioactivity of the lead compounds was performed. The ADME properties of the compounds were evaluated and the selected properties are linked to metabolism, cell permeation.

4 CONCLUSIONS

The importance of the study is due to the biological activity and ADMET study of these compounds determined by GC-MS Analysis. The prevailing compounds are Arabinopyranose, Myo-Inositol, D-Pinitol, Sorbopyranose, Talopyranose. The UV- VIS profile showed the peaks at 344.09 nm and 232.50 nm respectively. The results of FTIR analysis confirmed the presence of phenol, alkanes, aldehyde, secondary alcohol, amino acid and aromatic amines. The results of this study offer a platform of using *Saraca asoca*bark as herbal alternative for various diseases including diabetic, cardiovascular, cancer, gout, urolithiasis, urinary tract infection etc. However isolation of individual phytochemical constituents and subjecting it to biological activity will definitely give fruitful results.

G

5 ACKNOWLEDGMENTS

Authors are thankful to the Director of school of life sciences, Swami Ramanand Teerth Marathwada University, Nanded and the Indian Institute of Science (IISc), Bangalore for providing the instrumentation facilities.

REFERENCES

- [1] T. Athiralakshmy, A. Divyamol and P. Nisha, "Phytochemical screening of *Saraca asoca* and antimicrobial activity against bacterial species", Asian Journal of Plant Science and Research, 6(2)30- 36, 2016.
- [2] S. Misra andBhavaprakasa Nighantu.,"Indian Materica Medica" , Chaukambha Bharati Academy, 11th edition, Varanasi, pp. 500–501, 2007.
- [3] J. Harbone , "Phytochemical Methods". Champion and Hall Publishers, 2nd edition, London, pp. 84-196, 1984.
- [4] E. Frankel,"Nutritional benefits of flavonoids. International conference on food factors: Chemistry and Cancer Prevention", Hamamatsu, Japan, C6-2, 1995.
- [5] C. Kokate , A. Purohit and S. Gokhale ,"Practical Pharmacognosy", 2nd edition. Vallabh Prakashan, New Delhi, 466- 470, 2004.
- [6] V. Singleton, R. Orthofer, R. Lamuela-Raventos, "Analysis of total phenols and other oxidation substrates and antioxidants by means of Folin-Ciocalteau reagent", Methods Enzymol, 299:152-178, 1999.
- [7] R. Madaan, G. Bansal, S. Kumar andA. Sharma, "Estimation of total phenols and flavonoids in extracts of *Actaea spicata* roots and antioxidant activity studies",Indian Journal of Pharmaceutical Sciences,**73**(6): 666*-*669, 2011.
- [8] B. Sun, L. Concepcion, J. Ricardo Da Silva and I. Spranger, "Separation of grape and wine proanthocyanidins according to their degree of polymerization",Journal of Agricultural and Food Chemistry,46 1390–1396, 1998.
- [9] Bruna Medeiros-Neves, Francisco Maikon Corrêa de Barros, Gilsane Lino von Poser and Helder Ferreira Teixeira, "Quantification of Coumarins in Aqueous Extract of *Pterocaulon balansae* (Asteraceae) and Characterization of a New Compound", Molecules,20:18083-18094, 2015.
- [10]I. Koleva, T. Van Beek, J. Linssen, A. de Groot and L. Evstatieva, "Screening of plant extracts for antioxidant activity: a comparative study on three testing methods", Phytochemical Analysis,8-17, 2002.
- [11]C. Nisha , A. Kumar , B. Vimal , D. Bai andA. Pal,"Docking and ADMET prediction of few GSK-3 inhibitors divulges 6-bromoindirubin-3-oxime as a potential inhibitor", J. Mol. Grap. Model, 65 100- 107, 2016.
- [12]P. Chaturvedi, A. Ghatak and N. Desai,"Evaluvation of radical scavenging potential and total phenol content in *Woodfordia fruticosa* from different altitudes", Journal of Plant Biochemistry and Biotechnology, 21(1) 17-22, 2011.
- [13]K. Kumar and Devi Prasad,"Identification and comparison of biomolecules in medicinal plants of *Tephrosia tinctoria* and *Atylosia albicans* by using FTIR", Romanian J. Biophys, 21(1) 63-71, 2011.
- [14]Dukes. Phytochemical and Ethnobotanical Databases. Phytochemical and Ethnobotanical Databases. www.ars-gov/cgi-bin/duke/

[Gophane, *et al.,* **9(4): April, 2020] Impact Factor: 5.164 IC[™] Value: 3.00 CODEN: IJESS7 FIGURES** --- Absorbance at 415 nm --- Absorbance at 760 nm $12⁵$ y=0.011x+0.0636 1.0 $R^2 = 0.984$ $R^2 = 0.993$ 1.0 0.8

Fig.1: Standard calibration curve for total phenolic content for standard gallic acid , total flavonoids content for standard rutin , total proanthocyanidins content for standard catechin and total coumarin content for standard coumarin

http: // www.ijesrt.com**©** *International Journal of Engineering Sciences & Research Technology* [6]

ISSN: 2277-9655

Fig. 2: Percentage inhibition of DPPH free radical by crude extracts of Saraca asoca

Fig. 3: UV-Visible spectrum of ethanolic extract of S.asoca.

ISSN: 2277-9655

Fig. 4: FTIR spectrum of ethanolic extract of Saraca asoca

Fig. 5: GC-MS chromatogram of Saraca asoca ethanolic extract

[Gophane, *et al.,* **9(4): April, 2020] Impact Factor: 5.164**

TABLES

ISSN: 2277-9655 IC™ Value: 3.00 CODEN: IJESS7

 \circ

(+): Detected; (-): Not Detected

Values are expressed in mean±SD ; *P<*0.05.

Table III: IC⁵⁰ value of free radical scavenging activity by DPPH method

Values are expressed in mean±SD ; *P<*0.05.

Table IV: UV-VIS peak values of ethanolic extracts of Saraca asoca.

Table V: FTIR peak values and functional groups of Saraca asoca (ethanolic extract)

Table VI: Phytocomponents identified in the ethanolic extract of S.asoca using GC-MS.

(**Activity source: Dr. Duke's Phytochemical and Ethnobotanical Database) Table VIII: ADME profiles of the compounds

Compound 1: Arabinopyranose, tetrakis-O-(trimethylsilyl)-, à-D-

ISSN: 2277-9655

ADMET Predicted Profile --- Classification

Compound 2: Myo-Inositol, 1,2,3,4,5,6-hexakis-O-(trimethylsilyl)-

ADMET Predicted Profile --- Classification

ADMET Predicted Profile --- Regression

Compound 3: D-Pinotol, pentakis(trimethylsilyl) ether

ISSN: 2277-9655

ADMET Predicted Profile --- Classification

ADMET Predicted Profile --- Regression

[Gophane, *et al.,* **9(4): April, 2020] Impact Factor: 5.164**

ISSN: 2277-9655 IC™ Value: 3.00 CODEN: IJESS7

Compound 4:Sorbopyranose, 1,2,3,4,5-pentakis-O-(trimethylsilyl)-, L-

ADMET Predicted Profile --- Classification

http: // www.ijesrt.com**©** *International Journal of Engineering Sciences & Research Technology*

[16]

[Gophane, *et al.***, 9(4): April, 2020] Impact Factor: 5.164 Impact Factor: 5.164 ICTM Value: 3.00 CODEN: IJESS7**

ISSN: 2277-9655 IC™ Value: 3.00 CODEN: IJESS7

ADMET Predicted Profile --- Regression

Compound 5:Mannose, 2,3,4,5,6-pentakis-O-(trimethylsilyl)-, D-

ISSN: 2277-9655

ADMET Predicted Profile --- Classification

ADMET Predicted Profile --- Regression

ISSN: 2277-9655 IC™ Value: 3.00 CODEN: IJESS7

Compound 6: Mannose, 6-deoxy-2,3,4,5-tetrakis-O-(trimethylsilyl)

ADMET Predicted Profile --- Classification

ADMET Predicted Profile --- Regression

Compound 7: à-D-(+)-Talopyranose, pentakis(trimethylsilyl) ether

ADMET Predicted Profile --- Classification

ADMET Predicted Profile --- Regression

http: // www.ijesrt.com**©** *International Journal of Engineering Sciences & Research Technology* [21]

ISSN: 2277-9655